

Mass measurements at the FRS Ion Catcher and their application to nuclear structure studies

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Abstract

The FRS Ion Catcher (FRS-IC) enables precision experiments with thermalized exotic nuclei produced and separated in-flight in the Fragment Separator (FRS) at GSI. An unprecedented mass resolving power of almost 1000000 was achieved with the multiple-reflection time-of-flight mass-spectrometer (MR-TOF-MS) of the FRS-IC. Such high resolving power provides an exceptional opportunity to measure masses of short-lived exotic nuclei with down to few tens of keV uncertainty.

The results of experiments focused on various regions of the nuclide chart will be presented. These include:

- Several masses measured directly for the first time allowed to examine the evolution of two-neutron separation energy in a region above ^{208}Pb . Masses of ^{204}Au and ^{205}Au measured for the first time revealed deviations of up to 2.5σ compared to Atomic Mass Evaluation 2020 (AME20) extrapolated values. This resulted in a large change in the two-neutron separation energy at $N=126$.
- In the vicinity of ^{100}Sn , an isomeric state in ^{97}Ag was discovered for the first time using an MR-TOF-MS. This discovery was supported by mean-field calculations. The comparison of measured excitation energies of the $1/2^-$ isomers in odd indium isotopes $^{101-109}\text{In}$ with shell-model calculations showed the importance of including core excitations around ^{100}Sn .
- Direct mass measurements in the vicinity of ^{70}Br allowed to study a proton-neutron interaction strength in $N = Z$ region and provide a hint regarding the 500 keV discrepancy in the mass value of ^{70}Br , which impacts the F_t world average value for the superallowed $0^+ \rightarrow 0^+$ β -decays.
- A special mid-shell region above ^{100}Sn forms an island of nuclei with an α -decaying branch - Te, I, Xe, Cs, and Ba. Mass measurements of ^{114}I and ^{116}I allowed to estimate the α -decay partial half-life of ^{114}I with 2 orders of magnitude lower $\pm 1\sigma$ uncertainty and thus redefined the heaviest reported such an isotope of iodine which used to be ^{113}I .